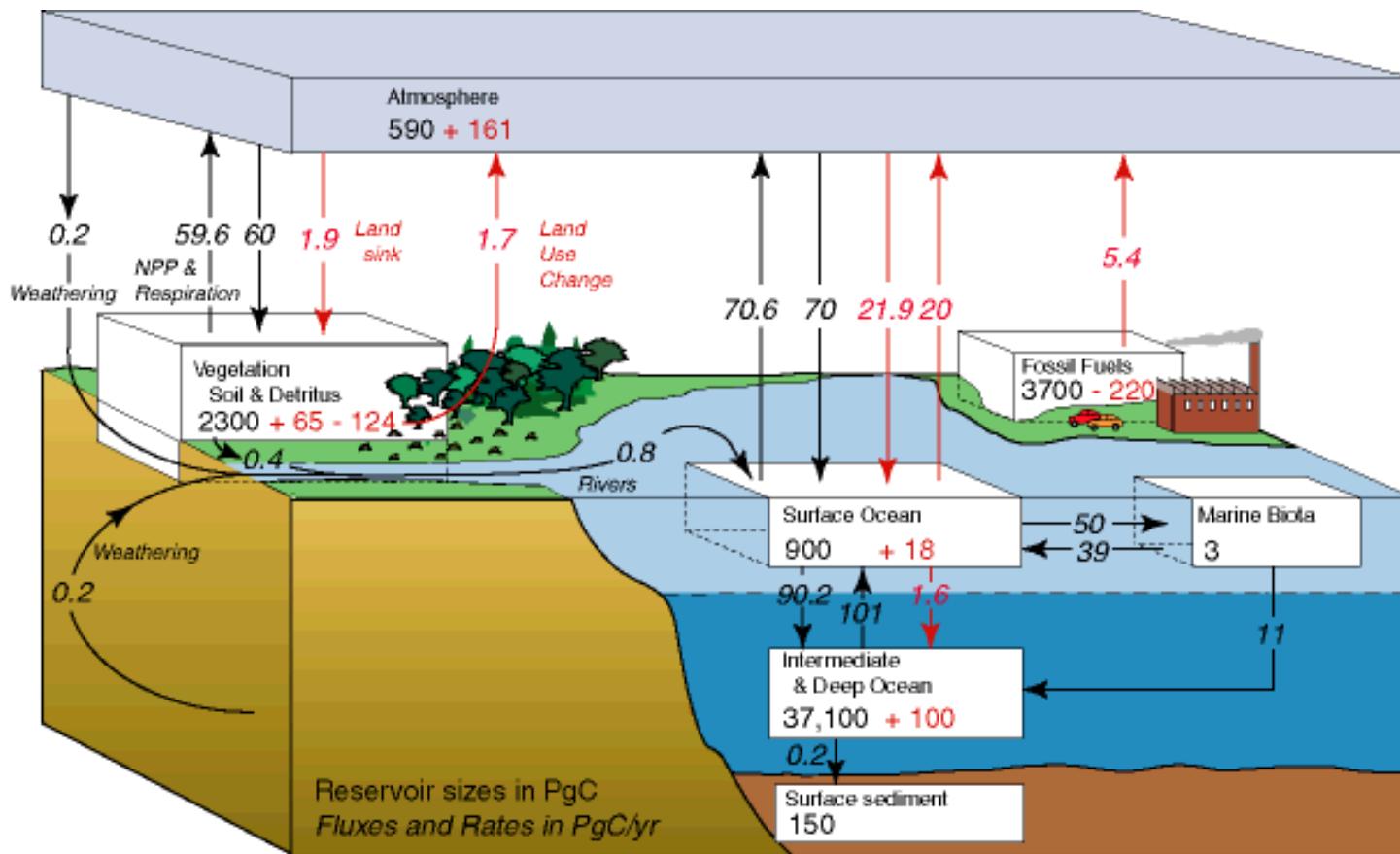


Tracing the present and past ocean carbon cycle

Matthieu Roy-Barman

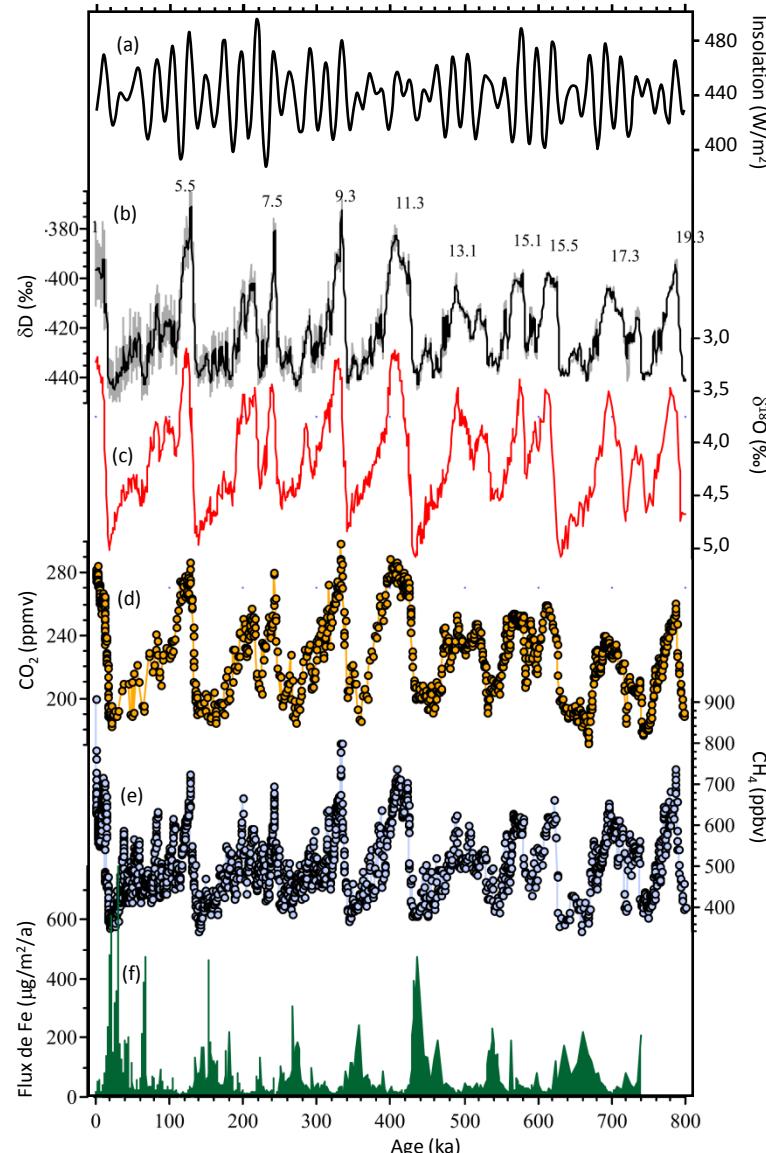
Laboratoire des sciences du climat et de l'environnement
Université de Versailles Saint Quentin

Global carbon cycle



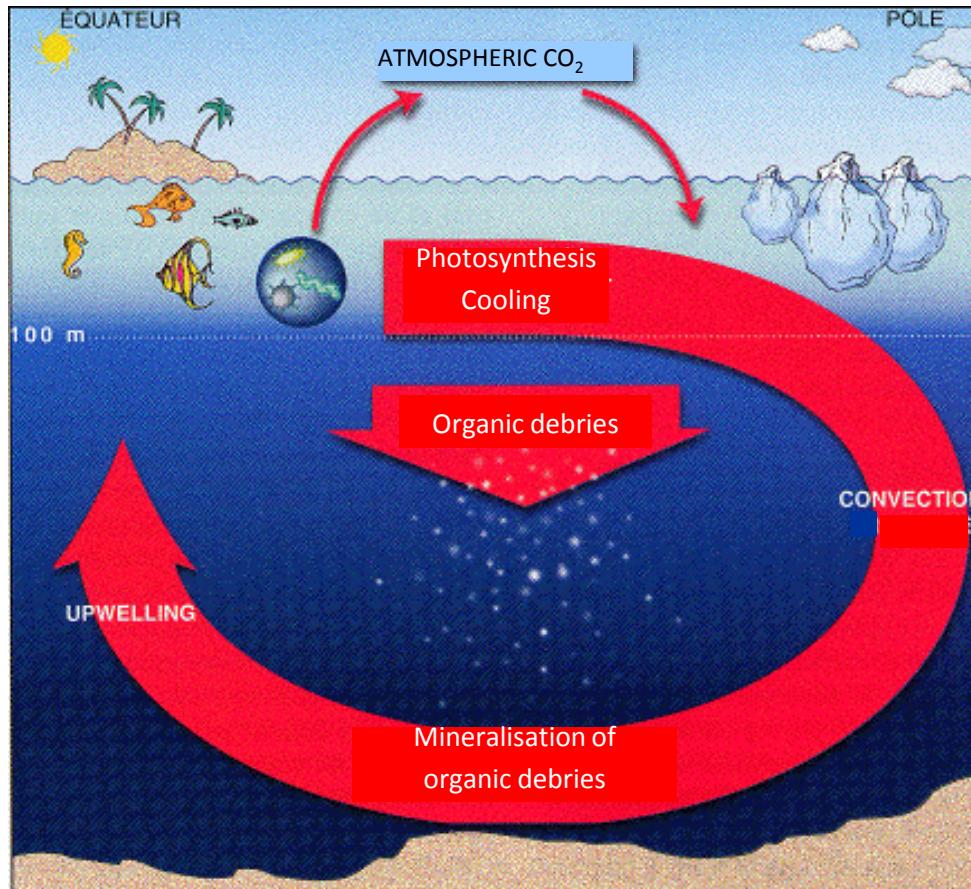
→ Ocean is the main C reservoir

Past carbon variability



Tracing the present and past ocean carbon cycle

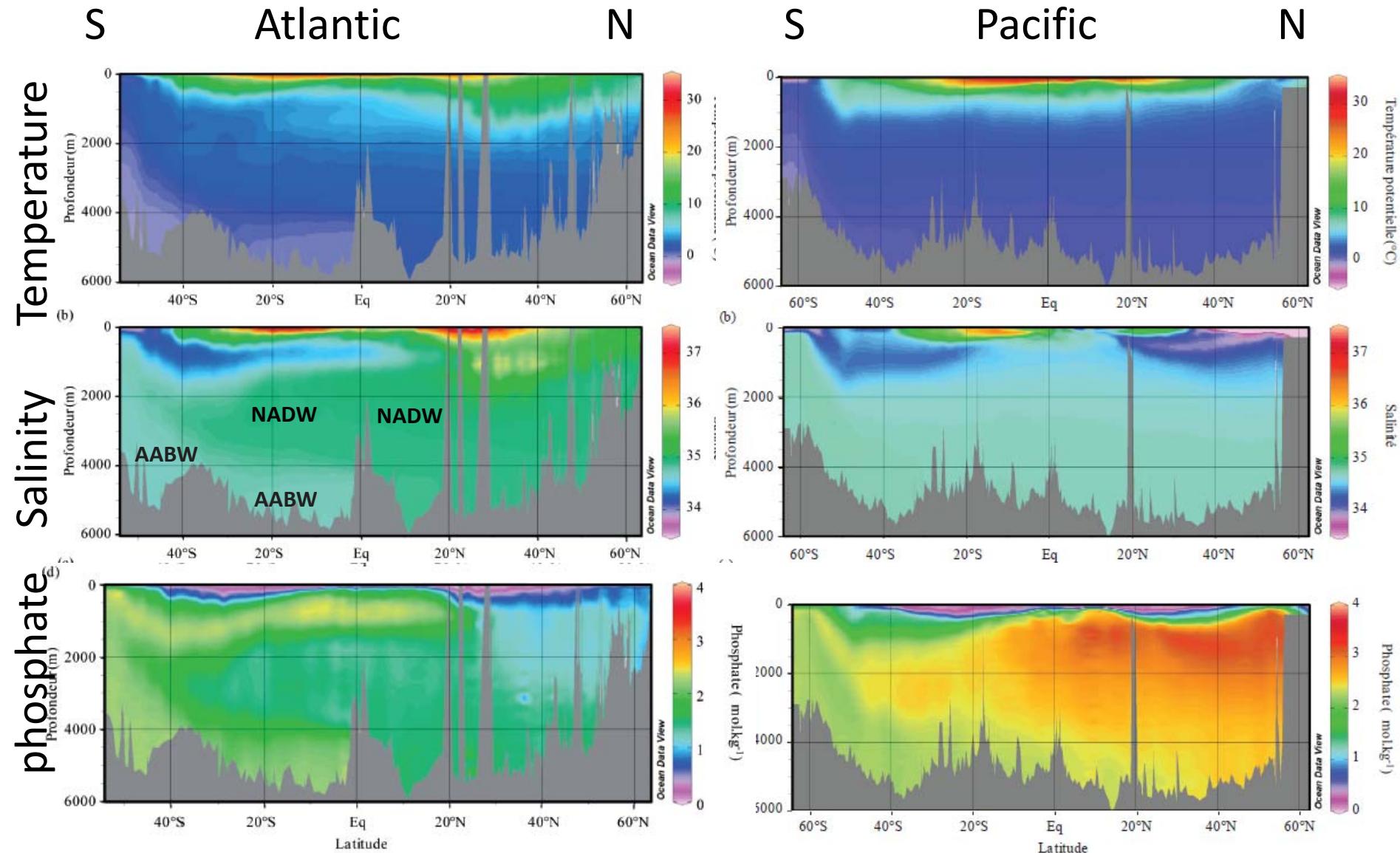
The oceanic carbon cycle



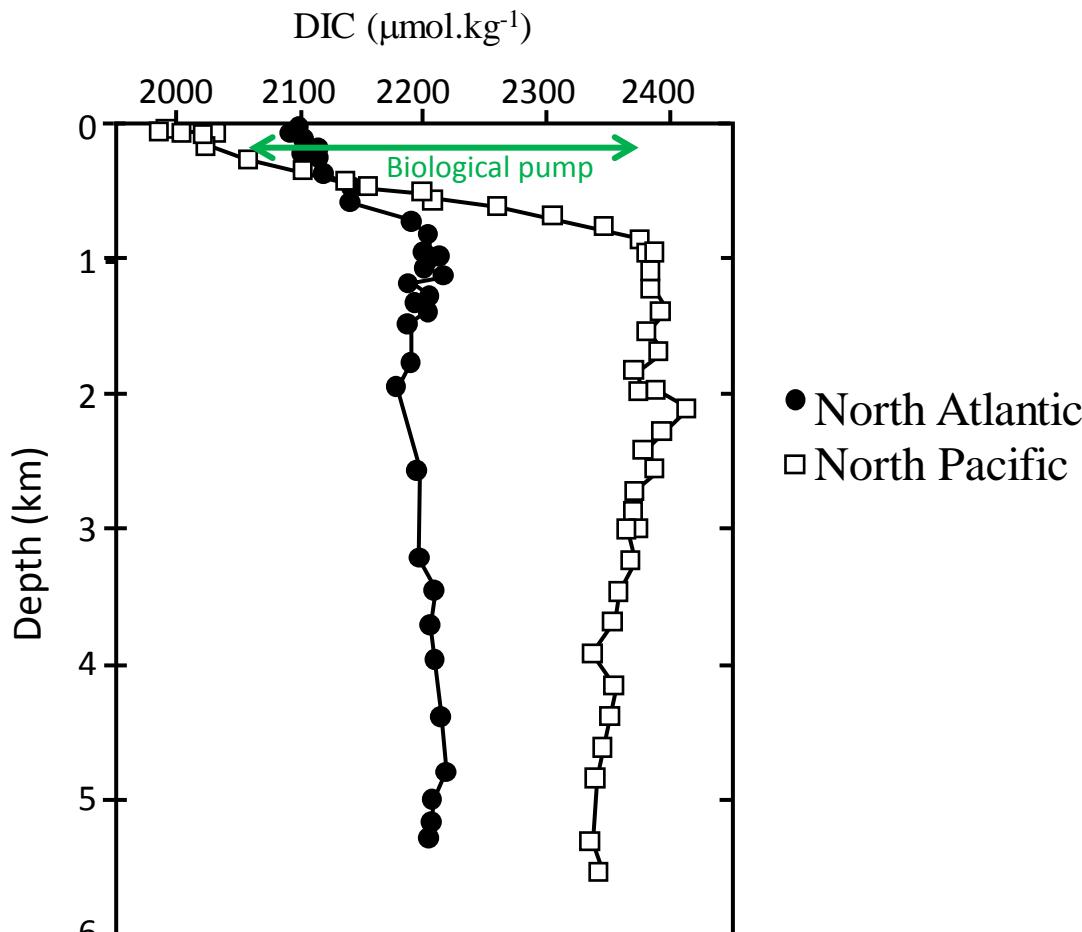
→ Requires constraints on dissolved and particulate carbon fluxes



Hydrographic sections

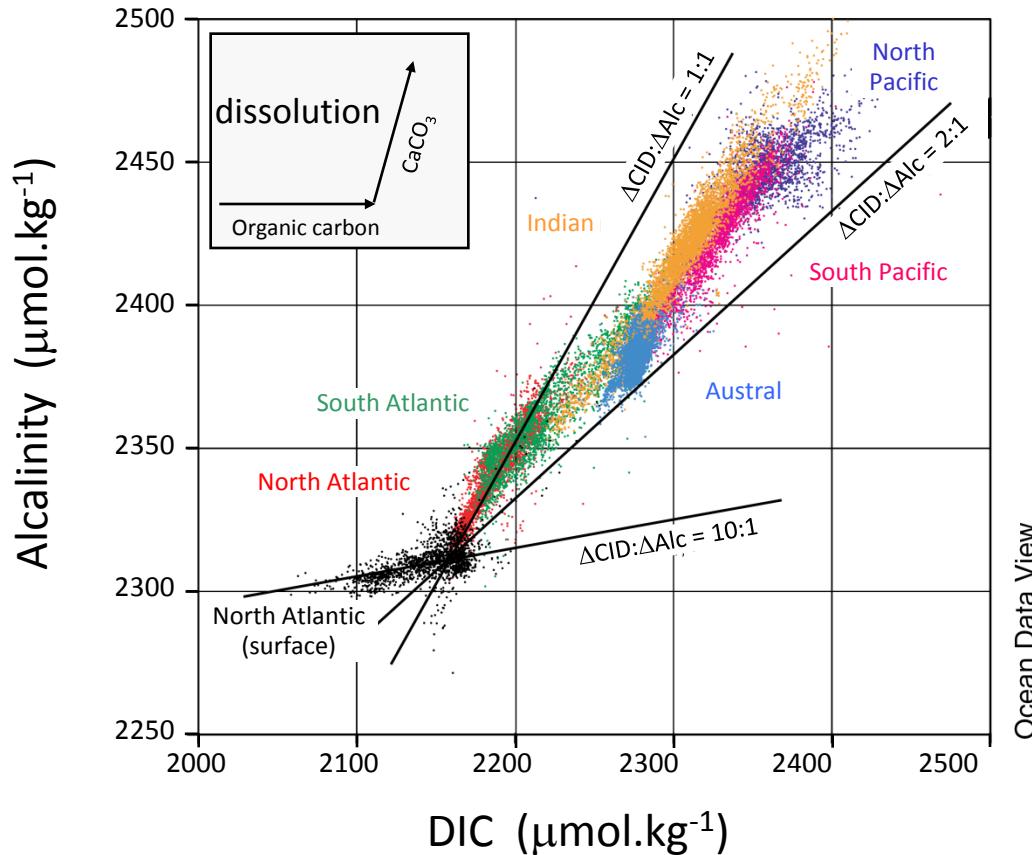


Dissolved inorganic carbon



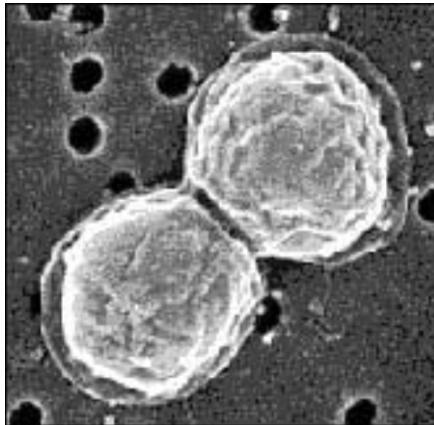
→ Requires constraints on dissolved and particulate carbon fluxes



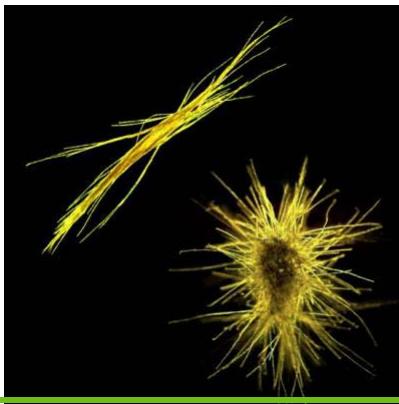


Photosynthetic bacteria

- Prochlorococcus (most abundant living being on earth)

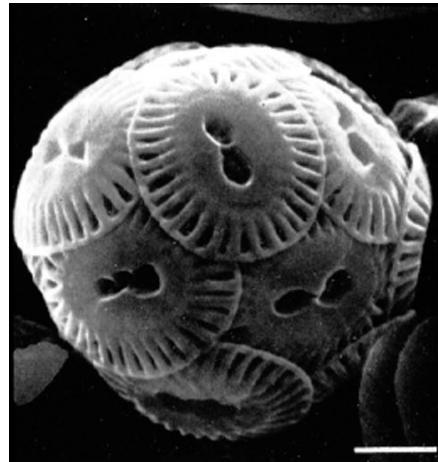


- Trichodesmium (N₂ fixing bacteria).

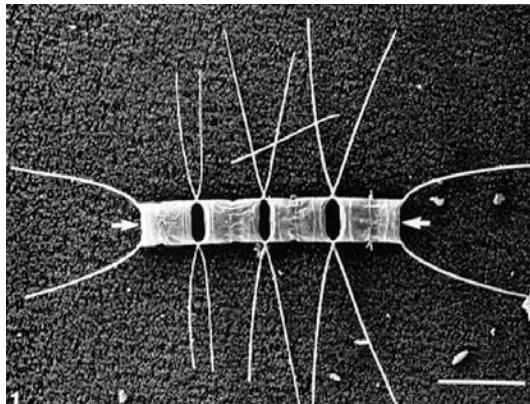


Photosynthetic eucariots

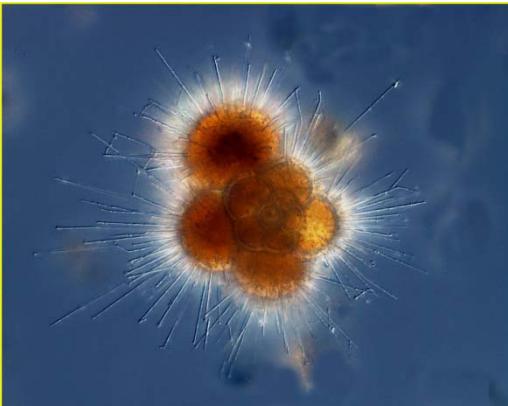
Coccolithophorids (calcium carbonate test)



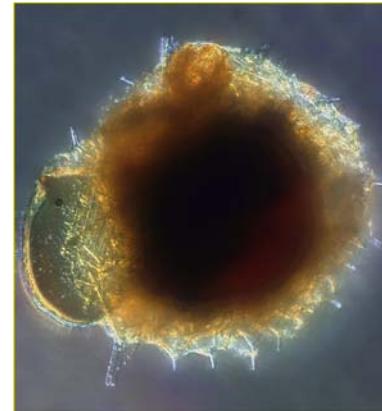
Diatoms (siliceous test)



Consumers



Foraminifera



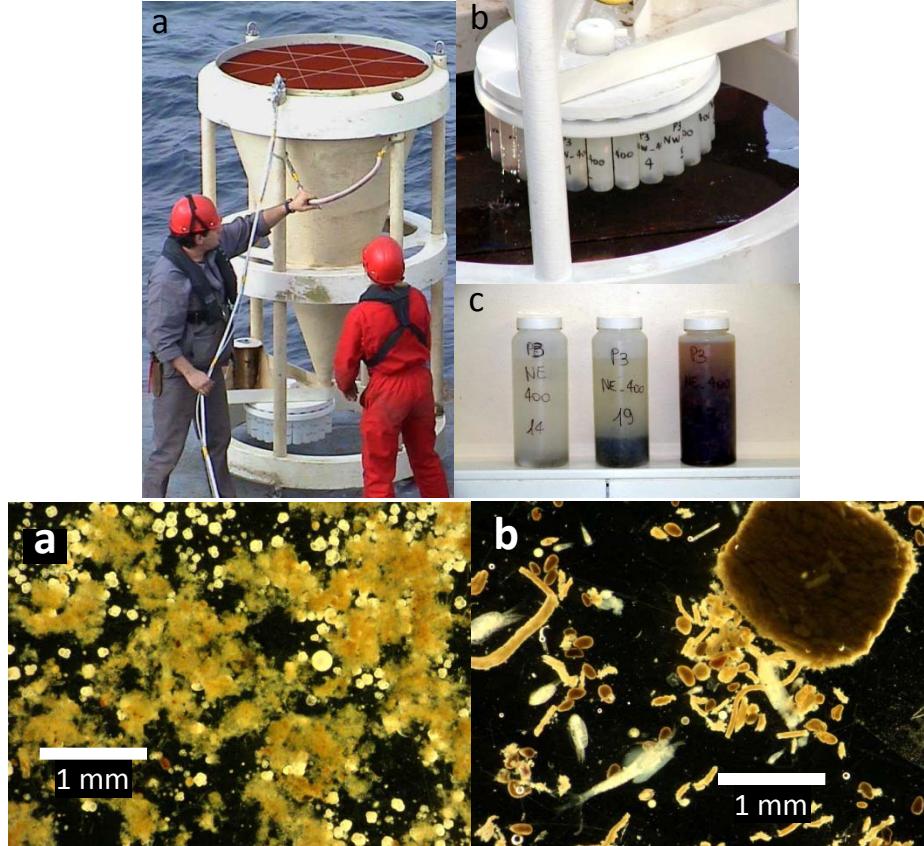
radiolarian eating a tintimid



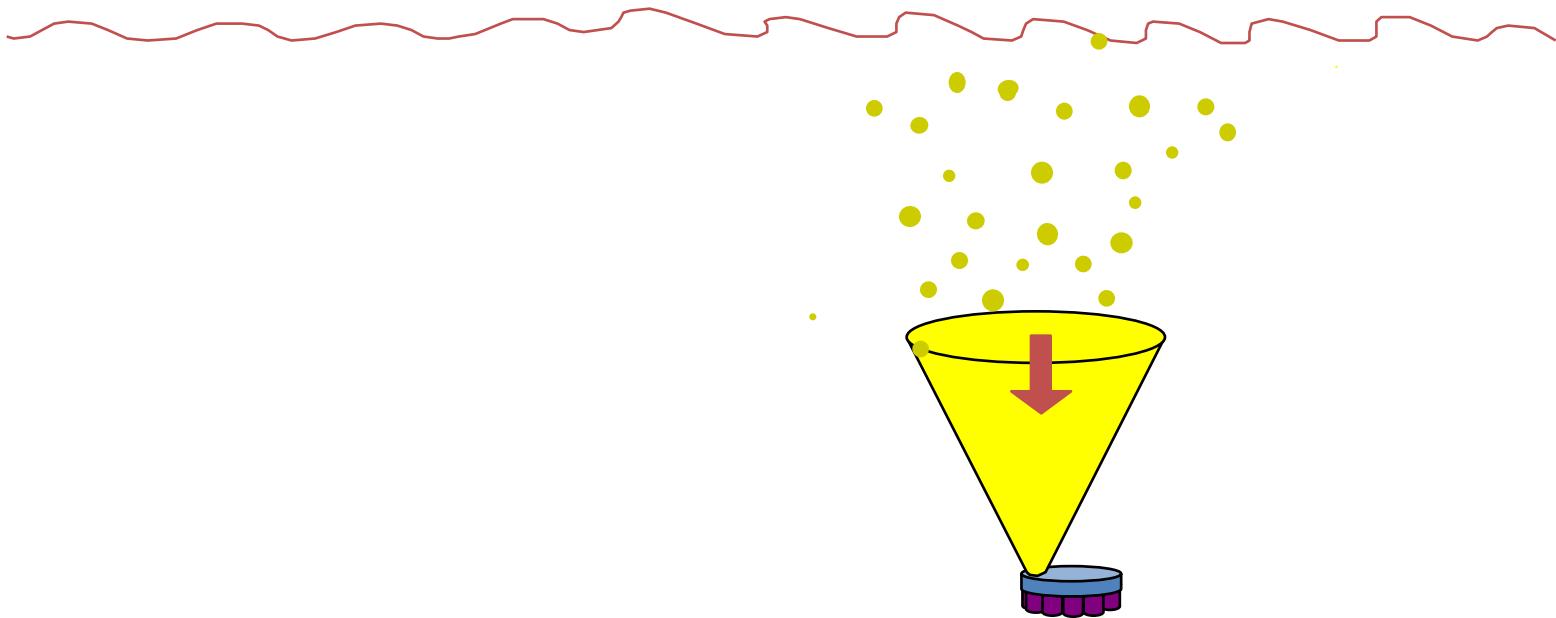
Seaweed



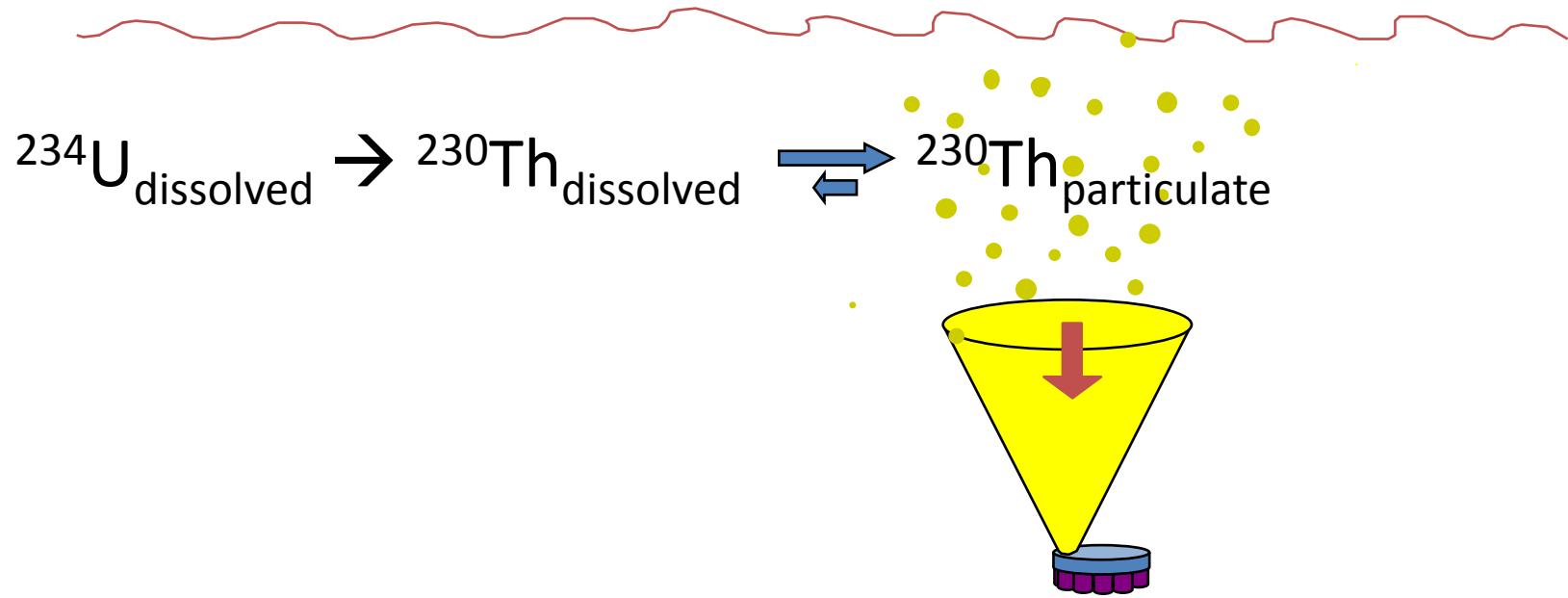
Measuring the biological pump



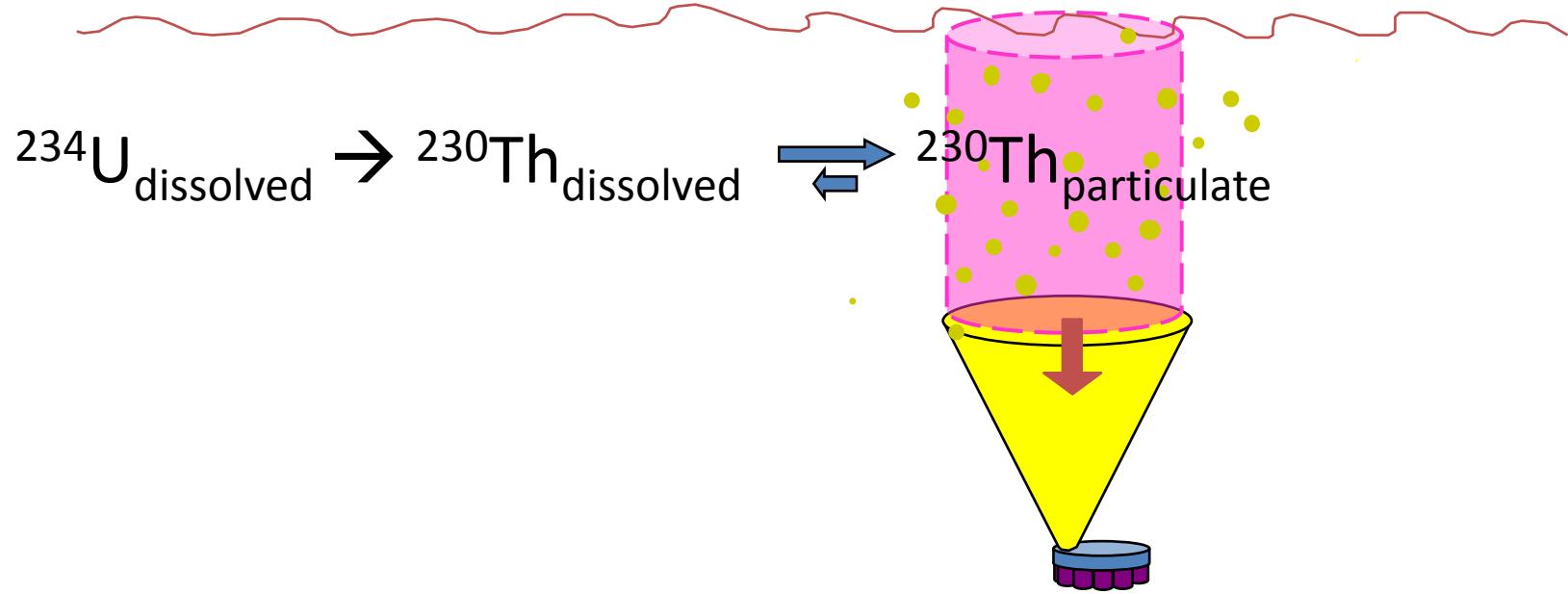
Estimation of particulate fluxes



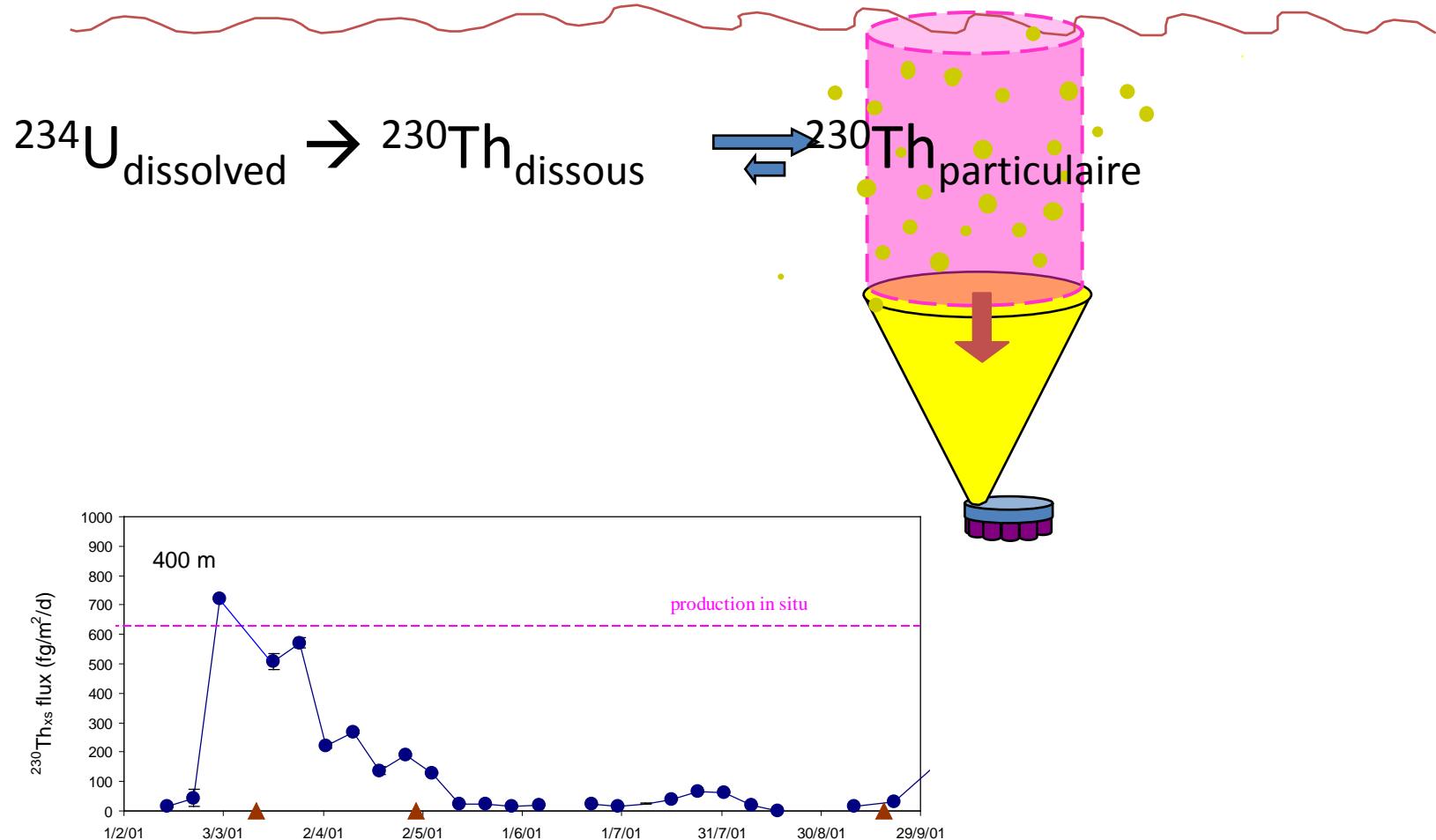
Estimation of particulate fluxes



Estimation of particulate fluxes



Estimation of particulate fluxes



Focussing/winnowing factor

$$\Psi_{i,j} = \text{MAR}_{i,j} \frac{\overset{\circ}{[\text{Th}]_{i,j}}}{P_{\text{Th}}(t_j - t_i)}$$

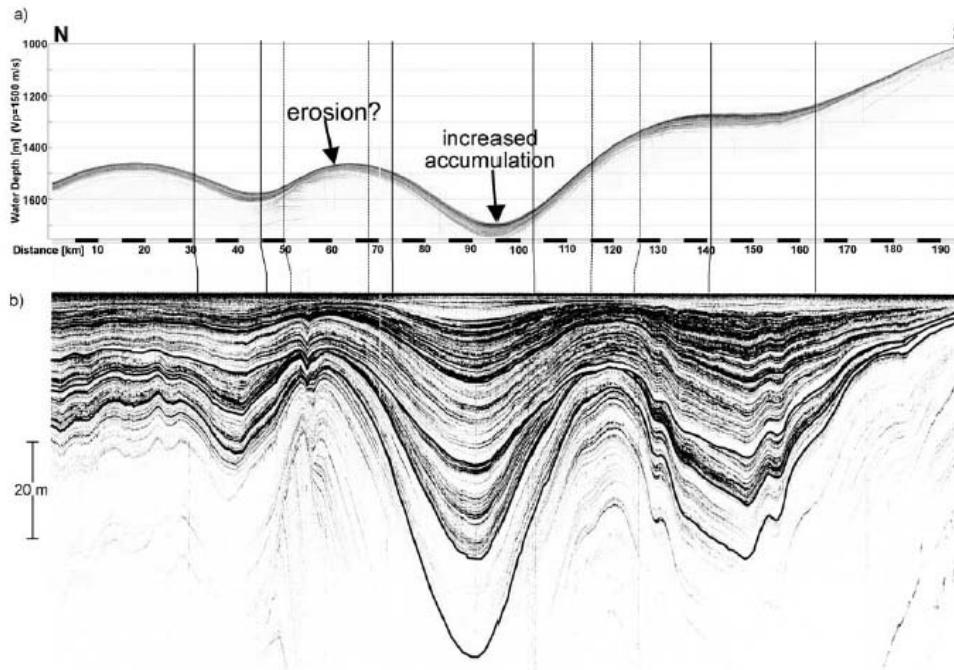
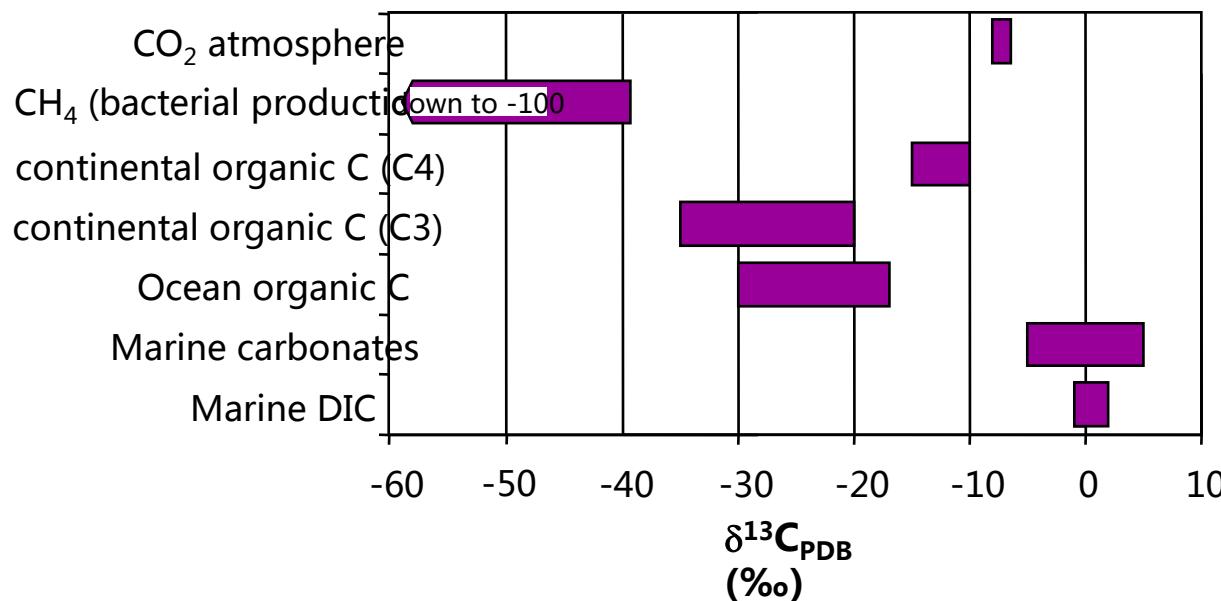


Figure 1. Echo sounder profile on a section parallel to the coast of Namibia showing evidence for sediment winnowing and pounding in mesoscale troughs [Mollenhauer et al., 2002].

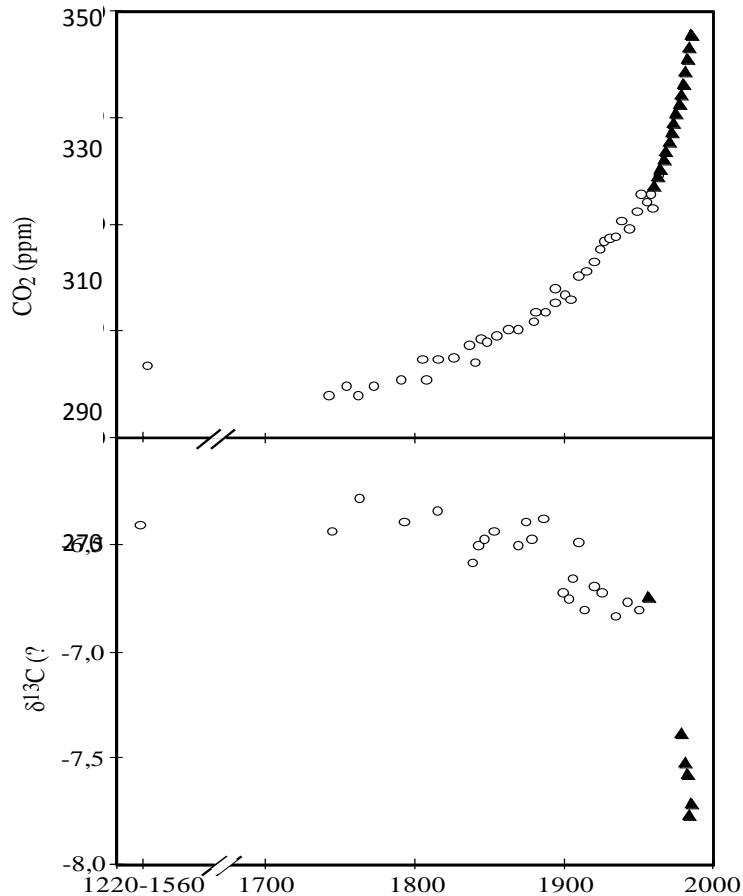
Normalisation to ^{230}Th allows quantifying sediment redistribution

carbon isotope fractionation

$$\delta^{13}\text{C}_{\text{sample/reference}} = \left\{ \frac{\left(\frac{^{13}\text{C}}{^{12}\text{C}}\right)_{\text{sample}}}{\left(\frac{^{13}\text{C}}{^{12}\text{C}}\right)_{\text{reference}}} - 1 \right\} \times 1000$$

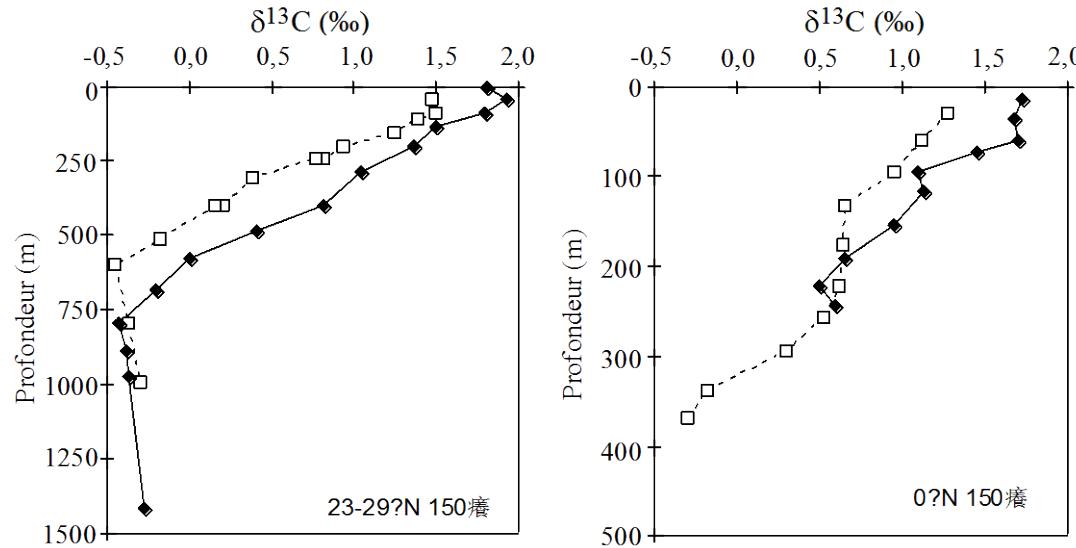


carbon isotope anthropic signature (Suess effect)



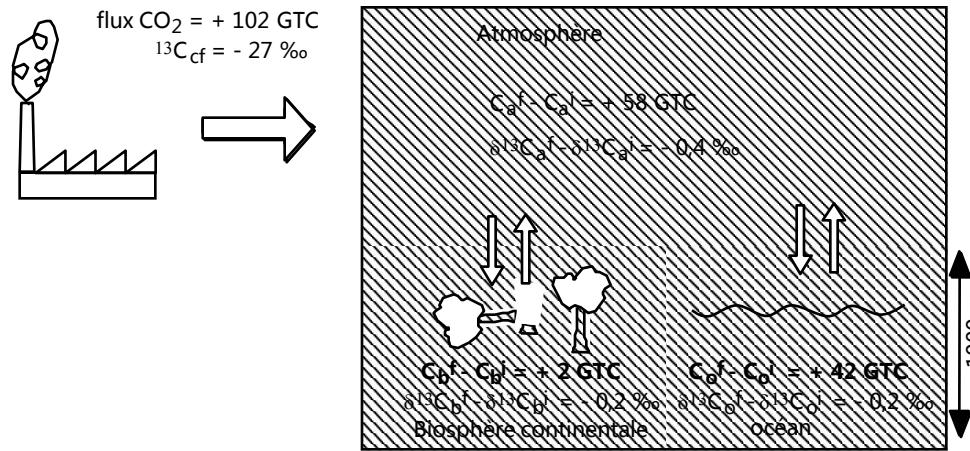
anthropic signature invading the ocean in 20 years

$\delta^{13}\text{C}$ measured 20 years appart



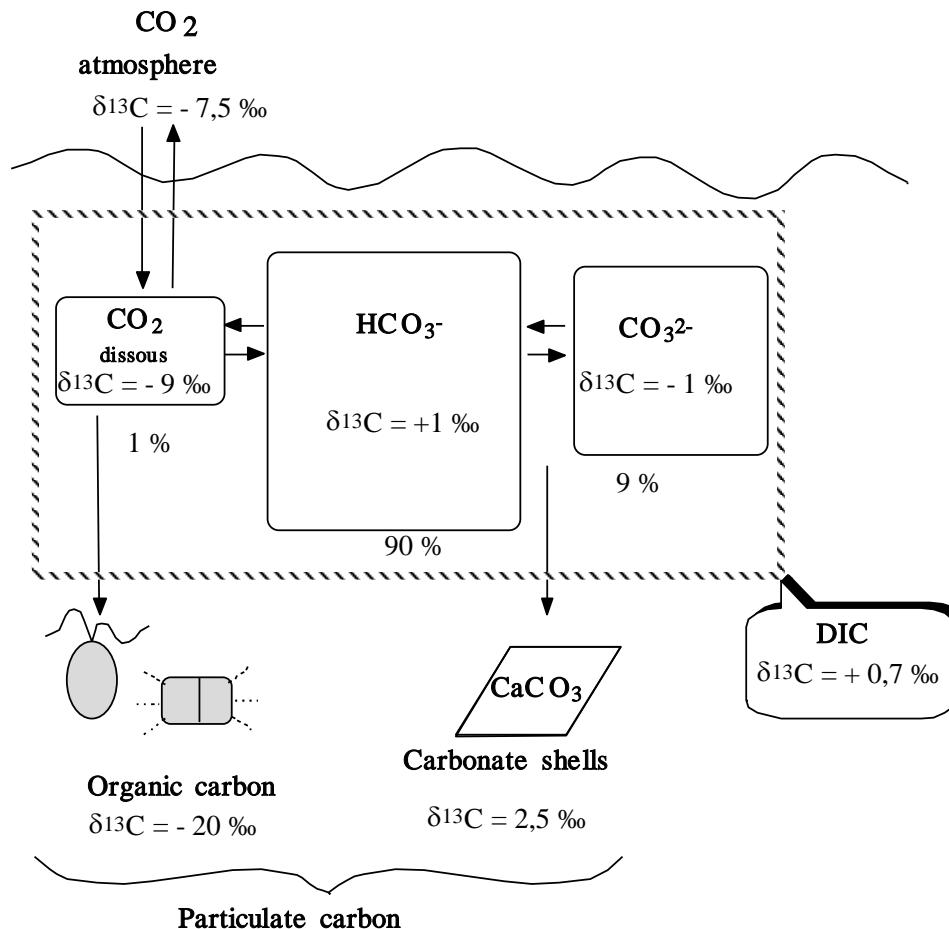
The first 500 m of is in contact with the atmosphere

Global 13C budget

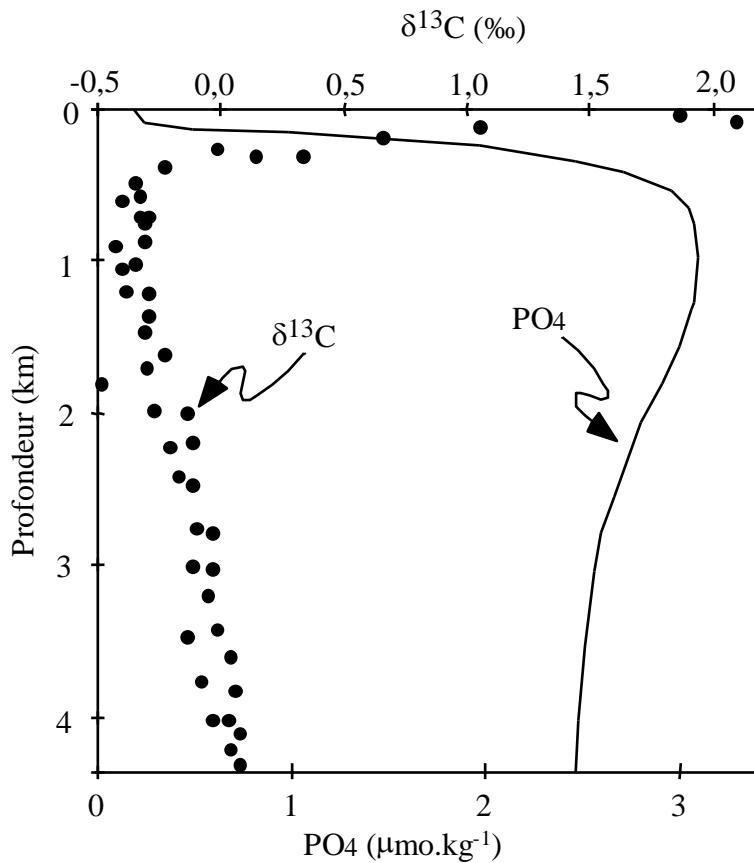


Ocean uptake > continental biosphere uptake

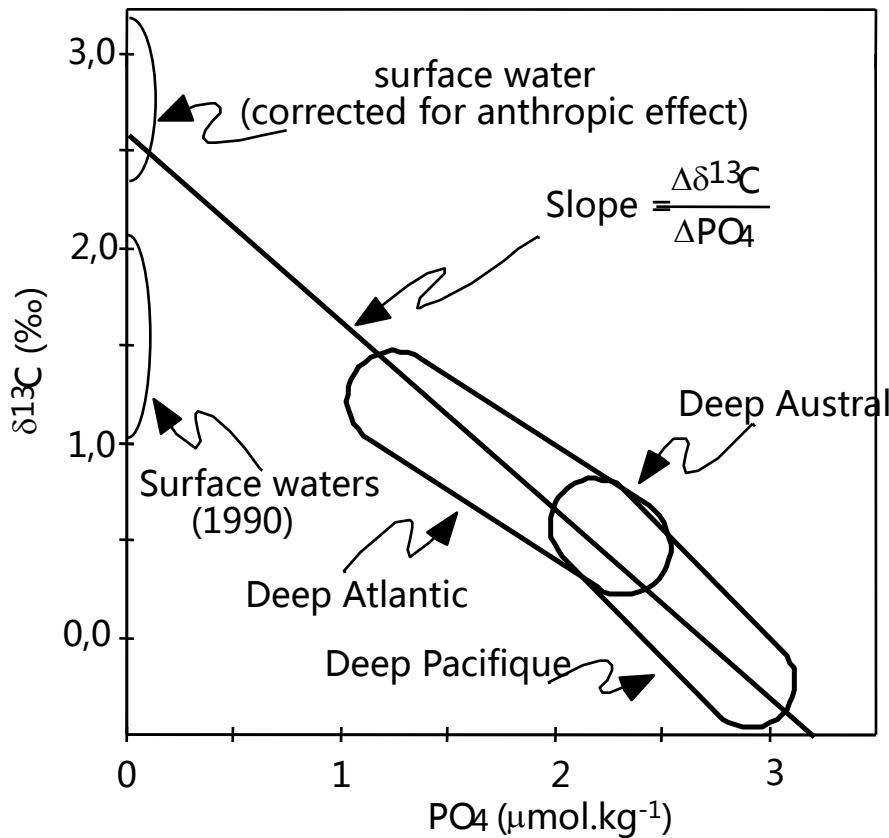
Carbon isotope fractionation in the ocean



Carbon isotope fractionation in the ocean

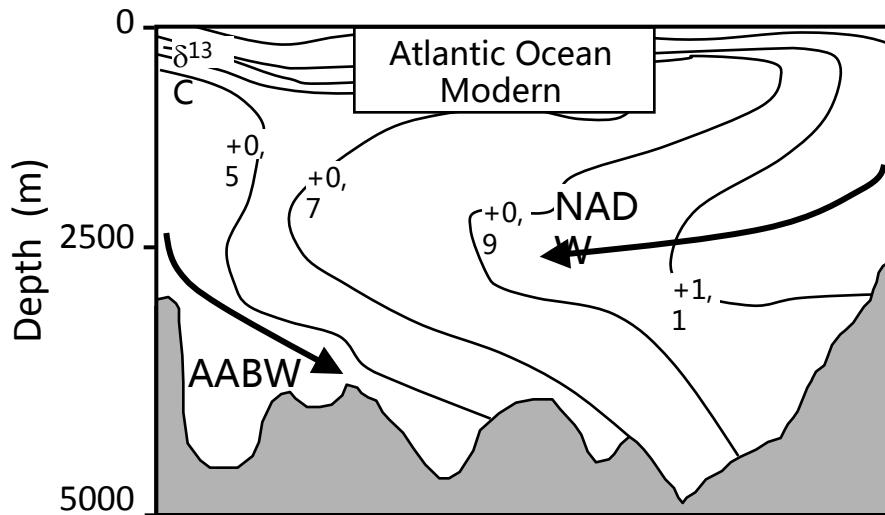


$\delta^{13}\text{C}$ for tracing watermasses

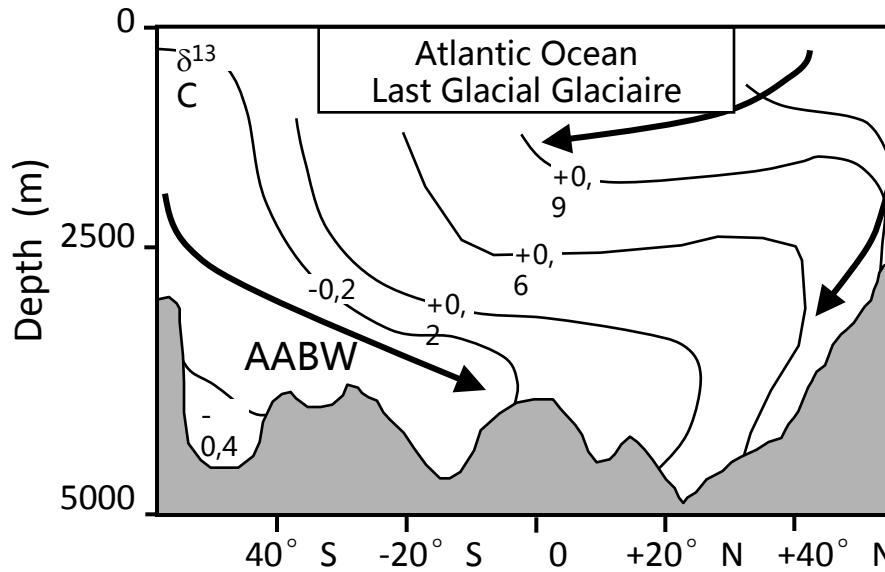


$\delta^{13}\text{C}$ in foraminifera

today

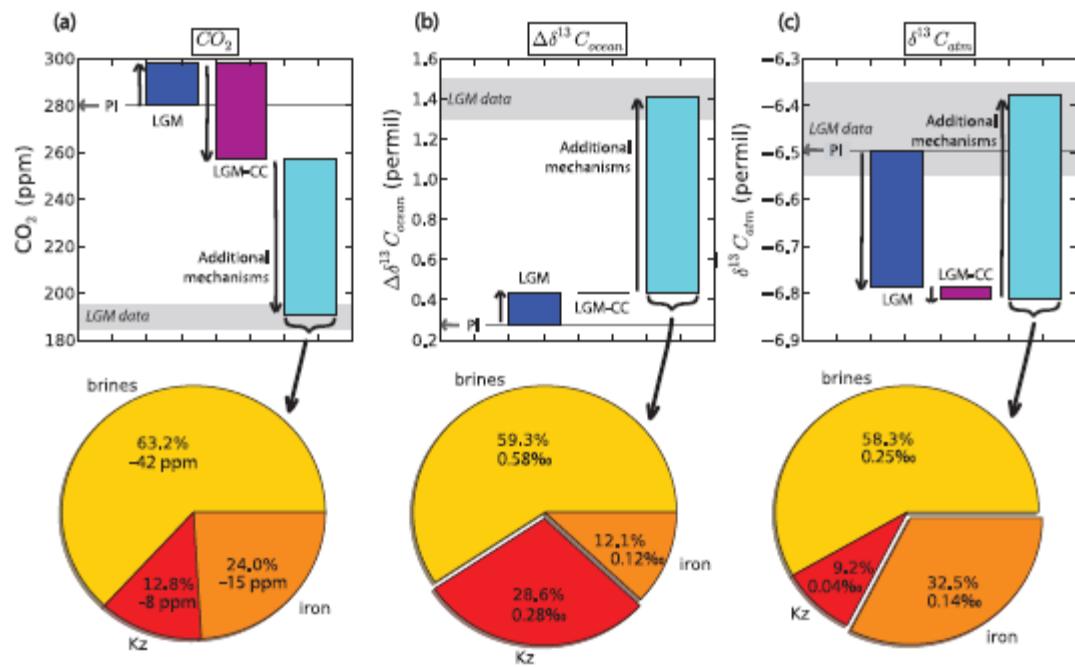
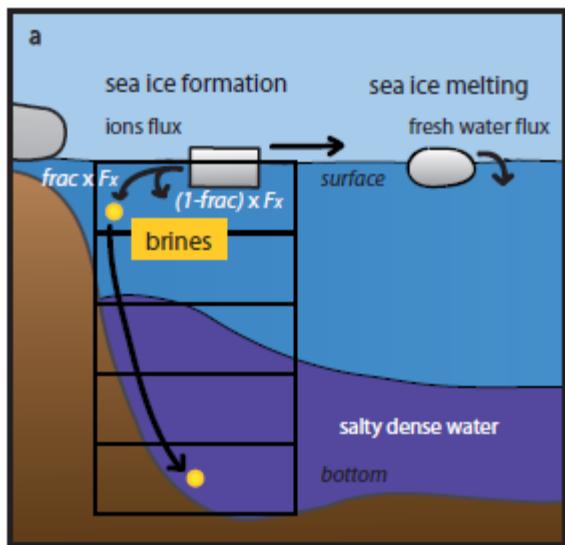


LGM



During LGM
storage of carbon
on the deep
ocean.

Balancing LGM carbon budget

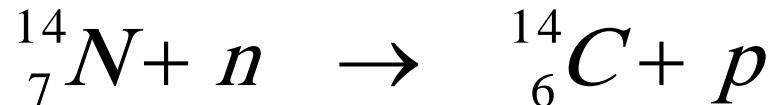


storage of carbon in the deep ocean explained by brine formation + iron fertilisation

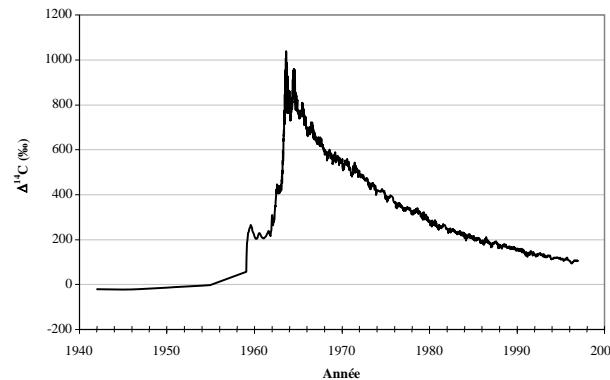
Bouttes et al. 2011

^{14}C production and fates

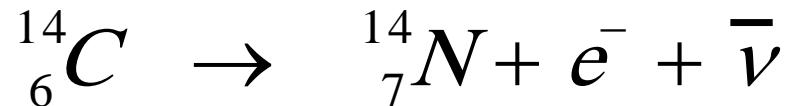
Natural production in the atmosphere



anthropic production in the atmosphere



Radioactive decay in superficial envelopes



^{14}C half life = 5700 y $\rightarrow \sim 12\%$ lost every century

Clock for the century-milennium timescale

^{14}C notations

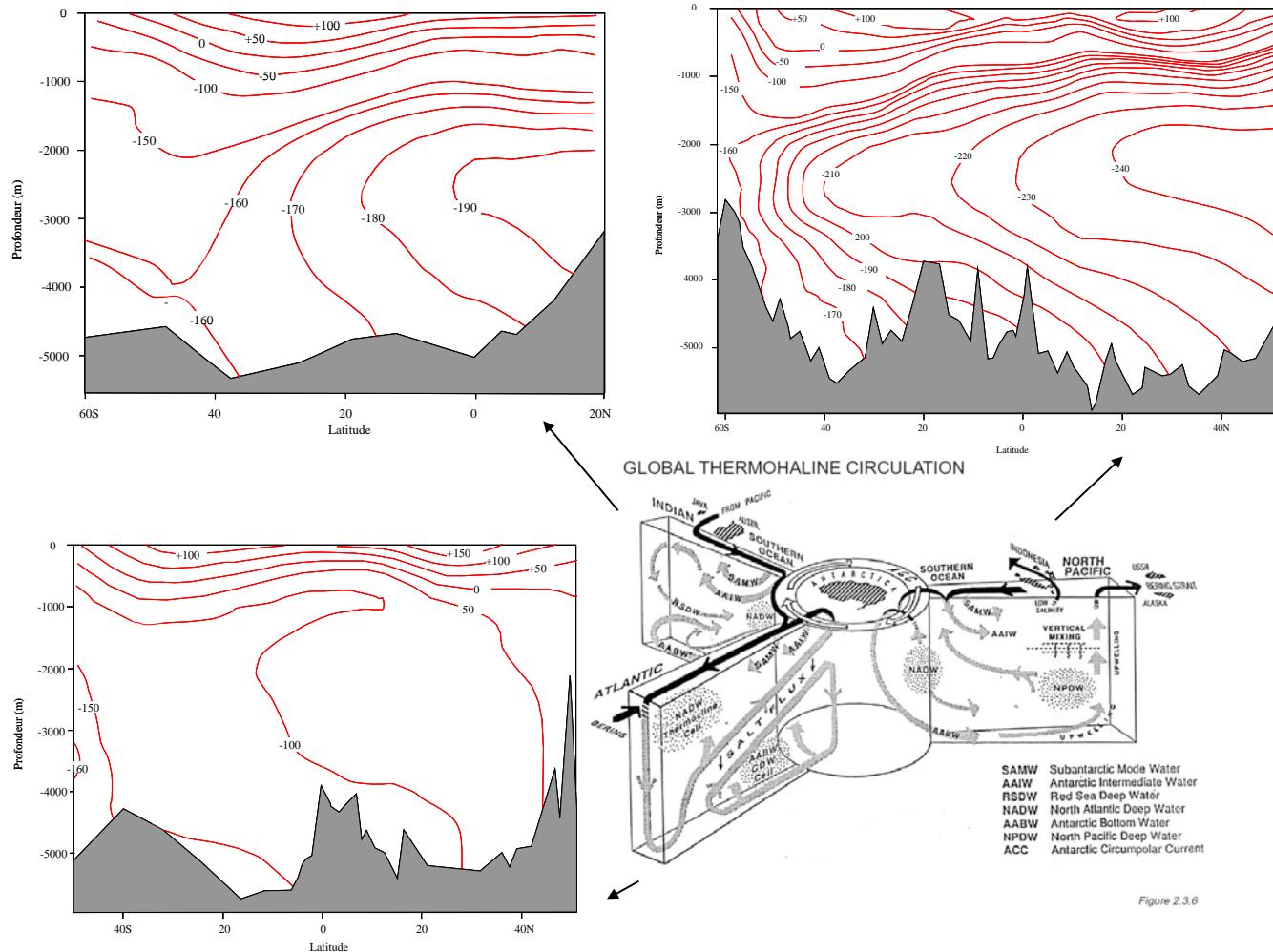
$$\Rightarrow \delta^{14}\text{C}_{\text{sample}} = \left\{ \frac{\left(^{14}\text{C}/^{12}\text{C} \right)_{\text{sample}}}{\left(^{14}\text{C}/^{12}\text{C} \right)_{\text{reference}}} - 1 \right\} \times 1000$$

avec $(^{14}\text{C}/^{12}\text{C})_{\text{référence}} = 1,2 \times 10^{-12}$

$$\Rightarrow \Delta^{14}\text{C}_{\text{sample}} = \left(\left(1 + \frac{\delta^{14}\text{C}_{\text{sample}}}{1000} \right) \times \left(1 + 2 \times \frac{25 - \delta^{13}\text{C}_{\text{sample}}}{1000} \right) - 1 \right) \times 1000$$

Correction of isotopic fractionnattion with $\delta^{13}\text{C}$

$\Delta^{14}\text{C}$ sections of the ocean



- Thermocline « invaded » by anthropic ^{14}C
- Aging of the deep water masses

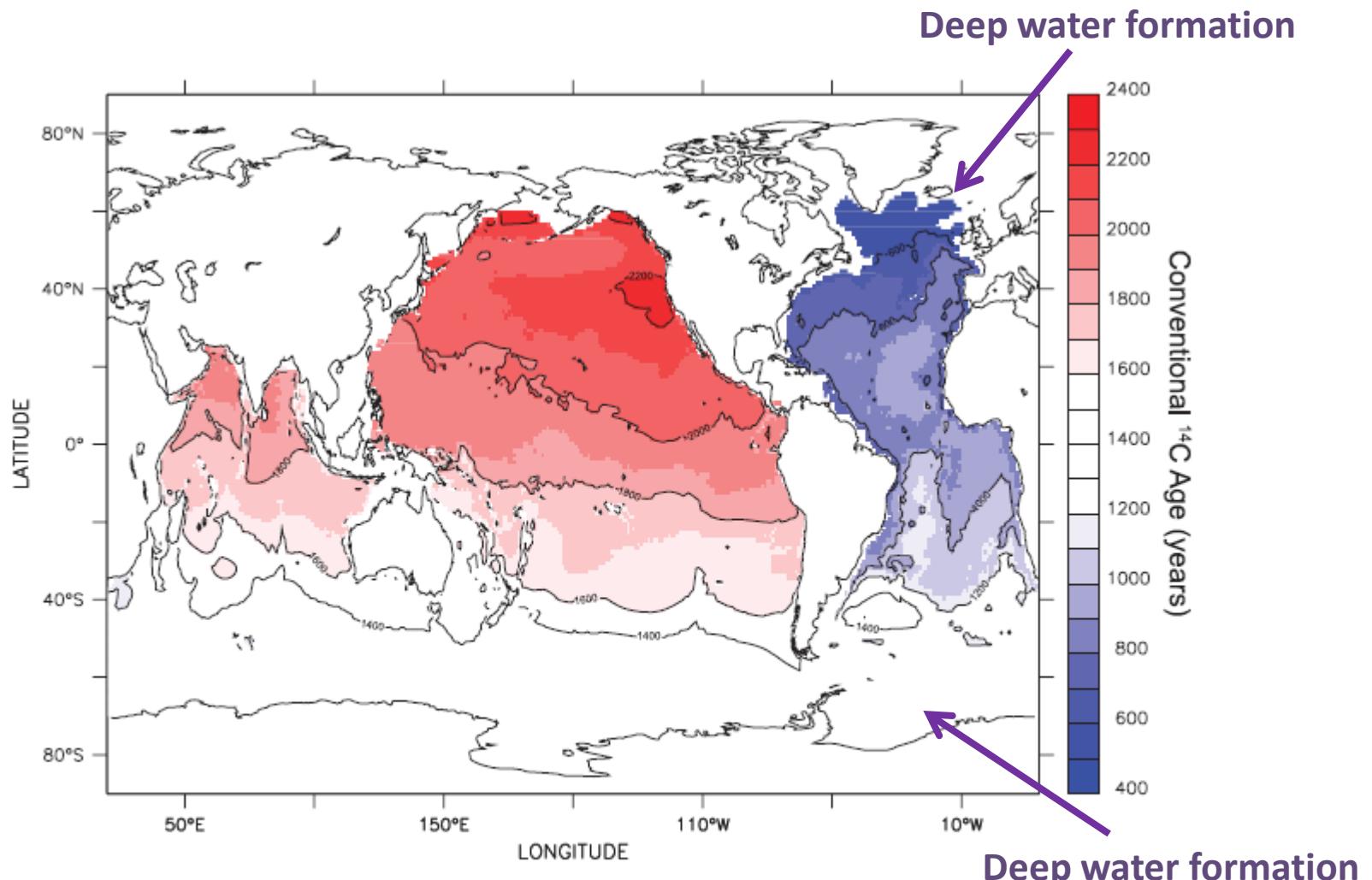
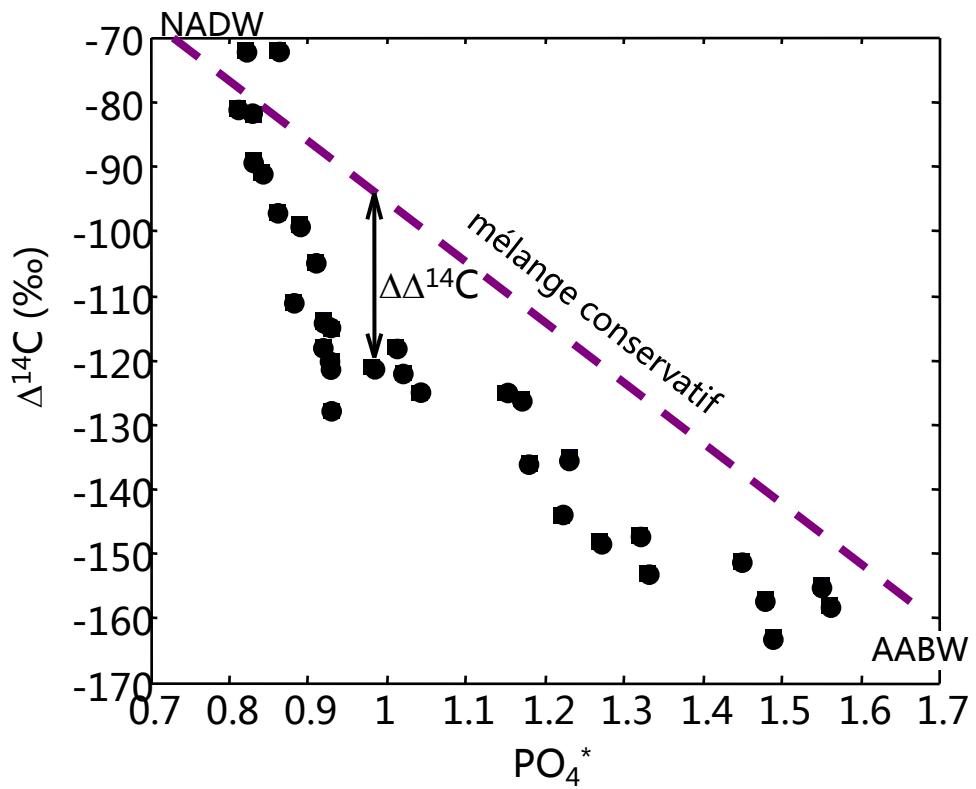
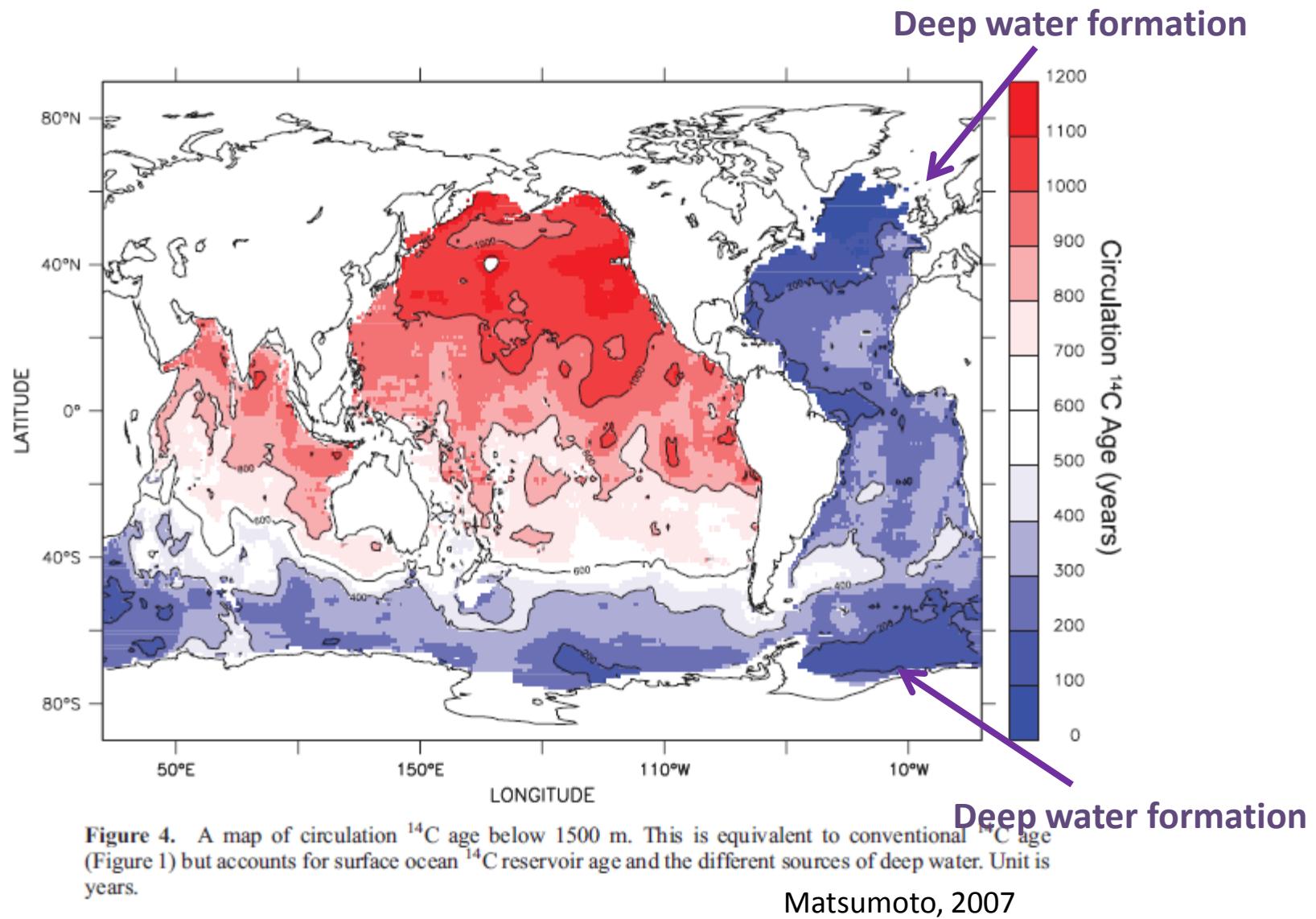


Figure 1. An objectively mapped conventional ^{14}C age of natural radiocarbon below 1500 m, following Matsumoto and Key [2004]. Unit is years.

Matsumoto, 2007



After Broecker, 1997



Comparing $\Delta^{14}\text{C}$ in surface and deep foraminifera

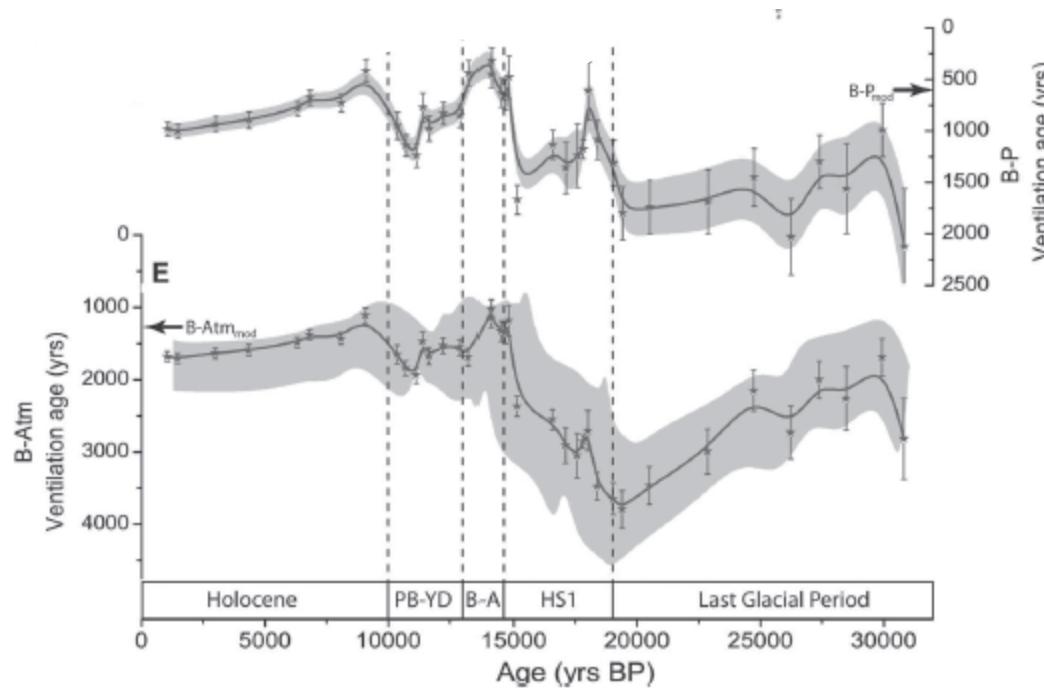
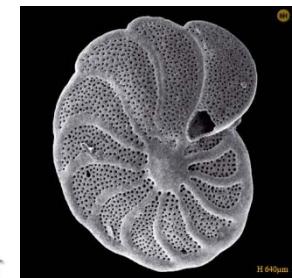
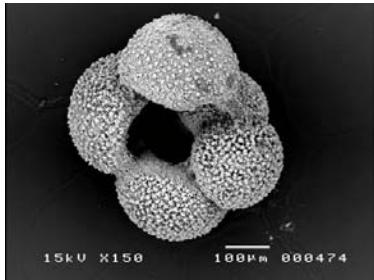


Fig. 2. Deep-water ventilation changes in the Atlantic sector of the Southern Ocean. (A) Atmospheric CO₂

→ Younger « ventilation » age in the southern Atlantic
today and at LGM

Neodymium isotope rationale

Neodymium is an insoluble element
(residence time = 500 y)



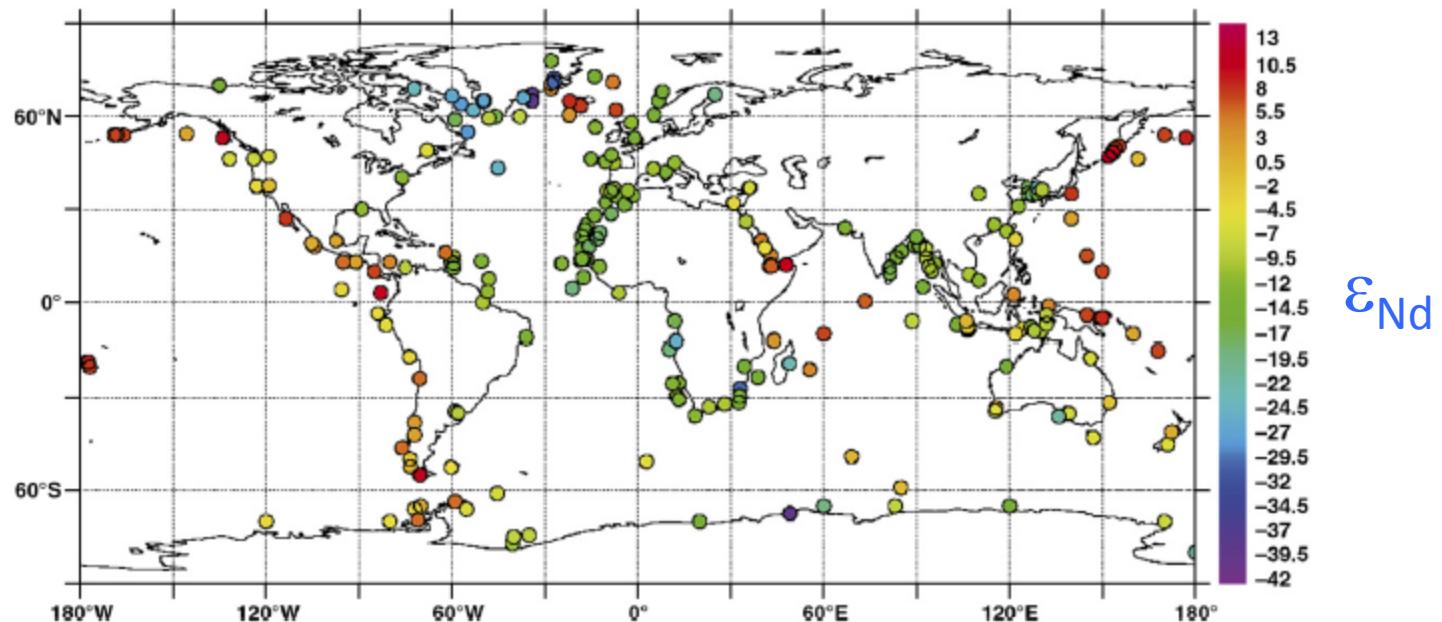
^{147}Sm half life = 106 Gy

→ very small changes of $^{143}\text{Nd}/^{144}\text{Nd}$

$$\varepsilon_{\text{Nd}} = \left(\frac{\left(\frac{^{143}\text{Nd}}{^{144}\text{Nd}} \right)_{\text{sample}} - 1}{\left(\frac{^{143}\text{Nd}}{^{144}\text{Nd}} \right)_{\text{CHUR}}} \right) \times 10^4$$

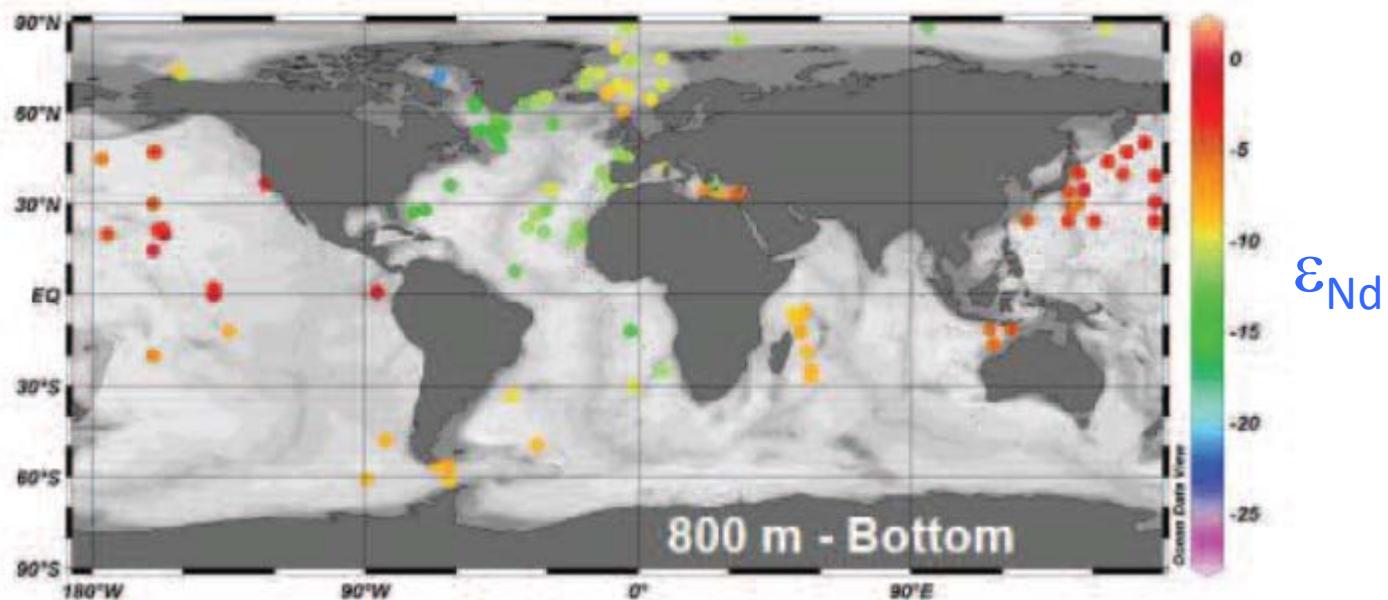


Rocks around the ocean



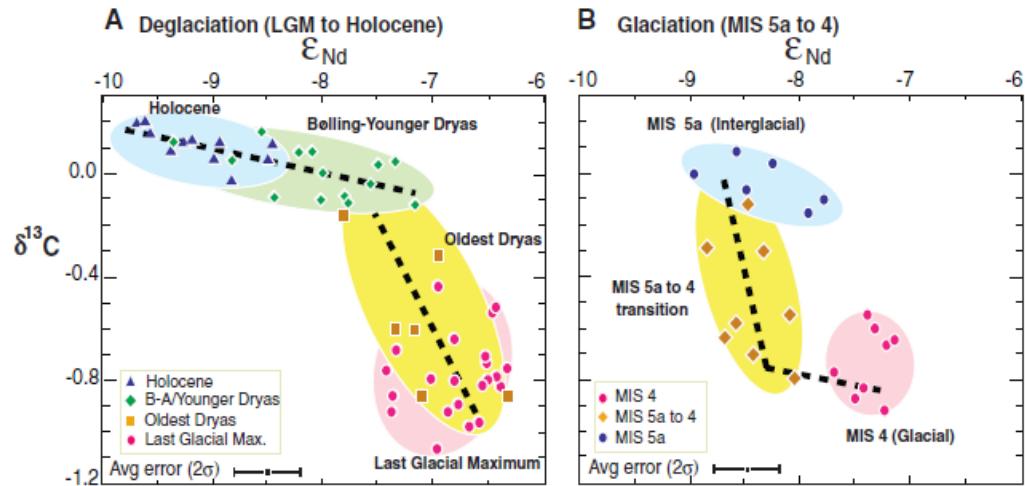
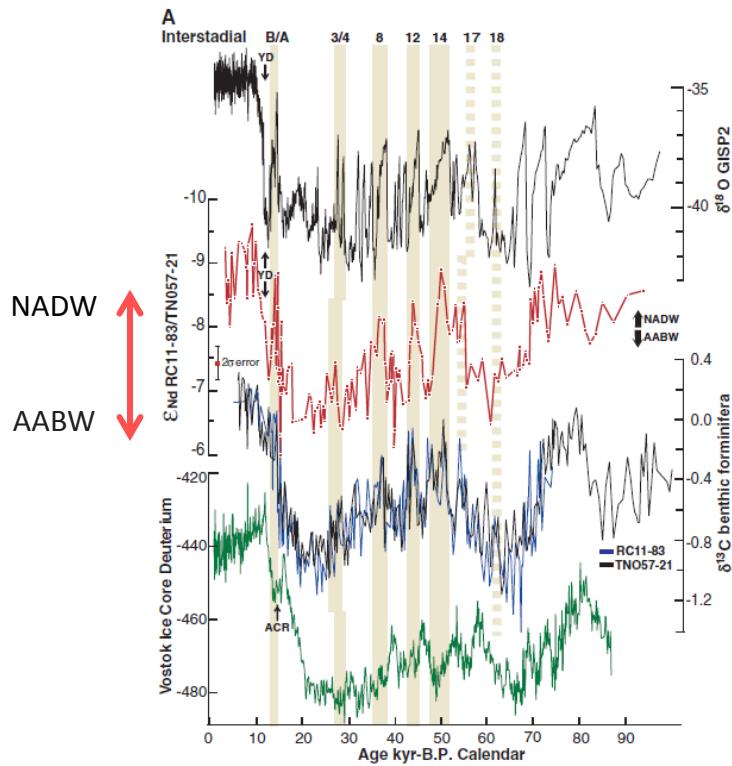
Jeandel et al., 2007

Water mass signatures



Jeandel et al., 2012

Neodymium isotopes -¹³C comparison

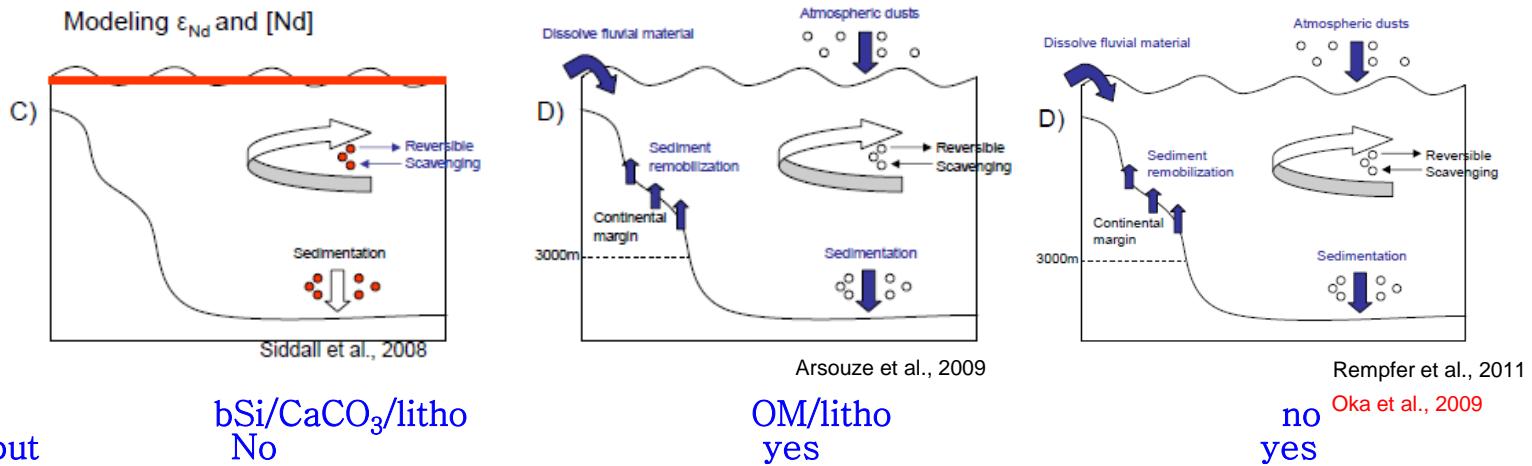
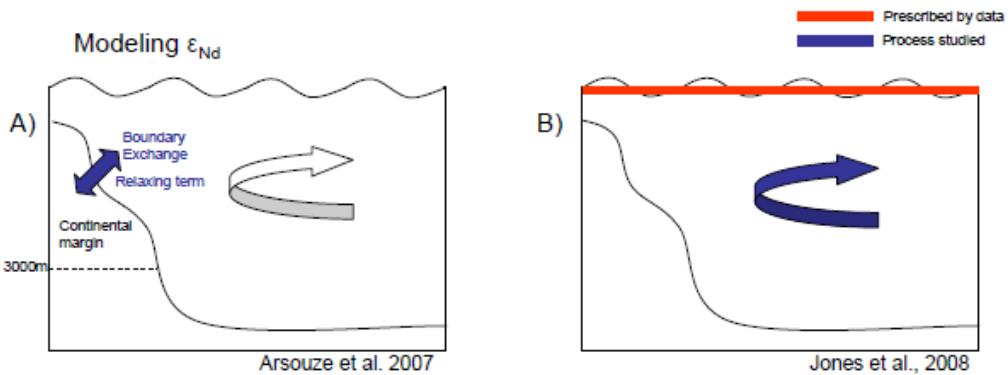


Piotrovski et al., 2005



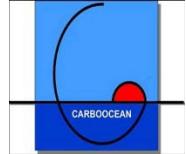
Nd modeling: the role of particles

No particles

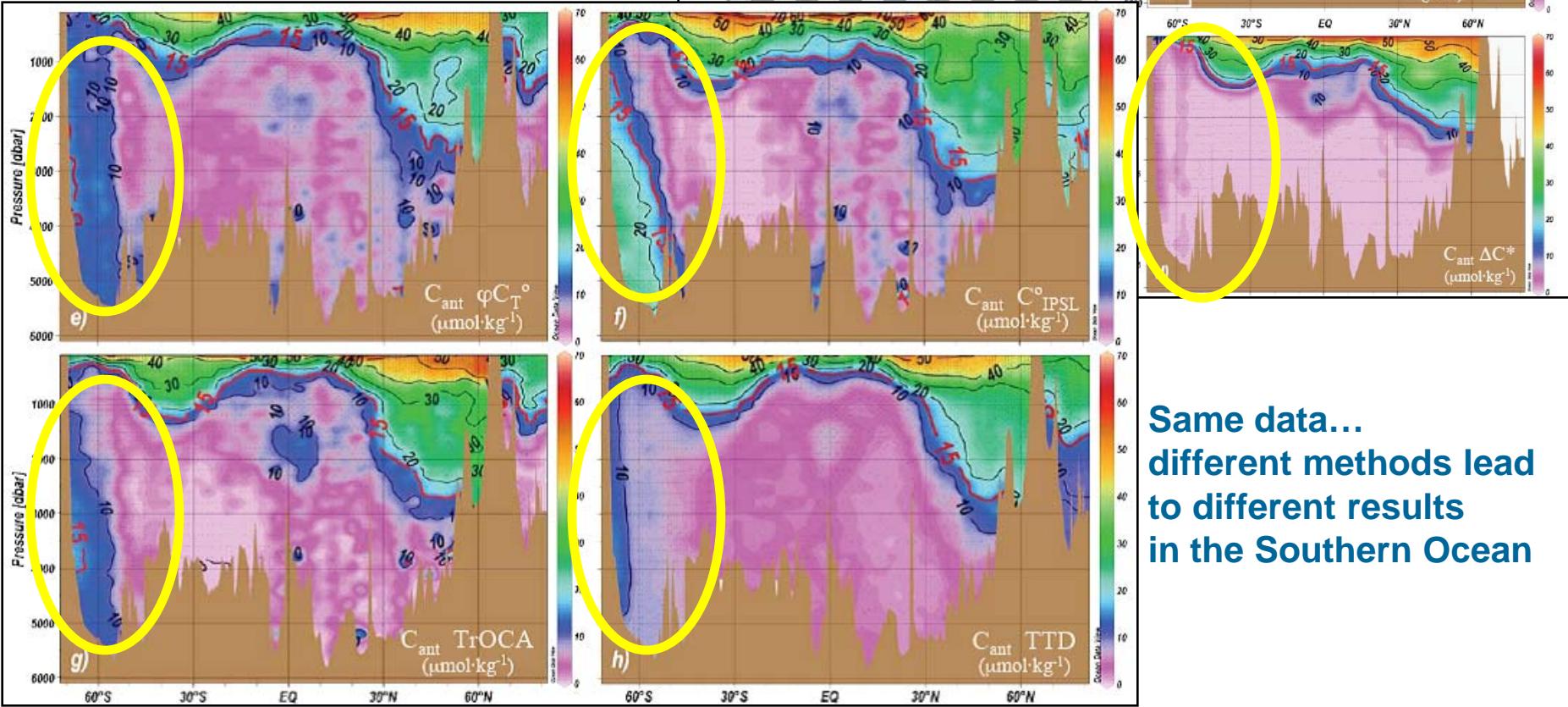
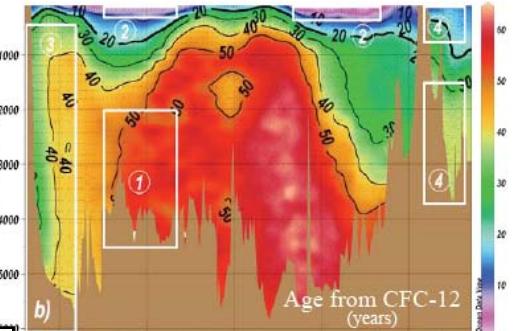
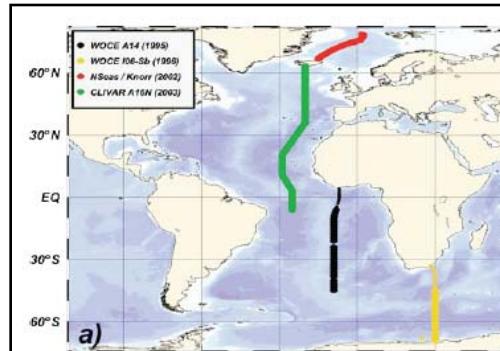


5 different models claim to reproduce « very well » the ε_{Nd}
in the ocean
upadapted from Arsouze et al., 2009

Atlantic: from the Arctic to the Southern Ocean



Vázquez-Rodríguez et al. (2009)

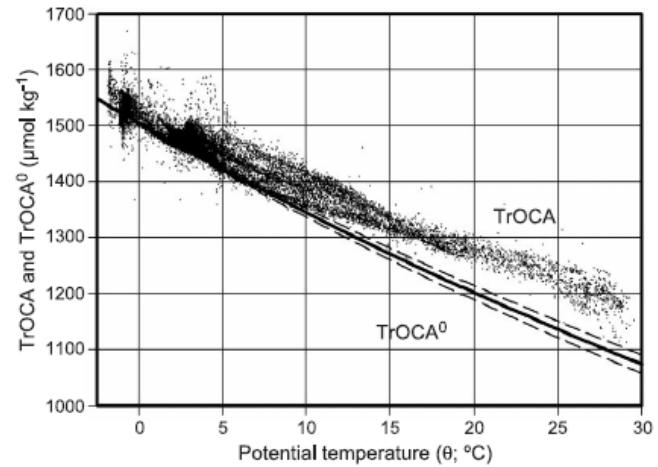


Troca Cook-book

$$\text{TrOCA} = \text{O}_2 + 1.2 \text{TCO}_2 - 0.6\text{TA}$$

$$\text{TrOCA}^0 = \text{O}_2^0 + 1.2 \text{TCO}_2^0 - 0.6\text{TA}^0$$

$$C_{\text{ANT}} = \text{TCO}_2 - \text{TCO}_2^0 = \frac{\text{TrOCA} - \text{TrOCA}^0}{1.2}$$



Touratier et al., 2004

Isotopic tracer are important at all scales:

- * source tracer
- * clock

trace elements are important:

- * specific informations